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## Emissions of C&D refuse in landfills: a European case

### **ABSTRACT:**

A field study was developed in a new landfill for refuse from construction and demolition (C&D) material recovery plants of small size (4 Ha.) in Europe, with the aim of evaluating the liquid and gas emissions in this type of facility at a large scale. It included characterization of the materials, monitoring leachate and gas quantity and composition. Besides thermometers, piezometers and sampling ports were placed in several points within the waste. This paper presents the data obtained for five years of the landfill life. The materials disposed were mainly made up of wood and concrete, similar to other C&D debris sites, but the amount of gypsum drywall (below 3% of the waste) was significantly smaller than other available studies, where percentages above 20% had been reported. Leachate contained typical C&D pollutants, such as different inorganic ions and metals, some of which exceeded other values reported in the literature (conductivity, ammonium, lead and arsenic). The small net precipitation in the area and the leachate recirculation into the landfill surface help explain these higher concentrations, thus highlighting the impact of liquid to solid (L/S) ratio on leachate characteristics. In contrast to previous studies, neither odor nuisances nor significant landfill gas over the surface were detected. However, gas samples taken from the landfill inside revealed sulfate reducing and methanogenic activity.

### **KEY WORDS:**

*C&D waste, leachate, landfill gas, recirculation, instrumentation*

## 1. INTRODUCTION

The construction industry has a large economic and social role in numerous countries. Traditionally, its development entails an increase in the production of construction and demolition (C&D) waste, with the subsequent loss of resources and environmental impact. In order to avoid these effects, specific standards have been promulgated recently to regulate the generation and management of this waste, promoting recycling and seeking the reduction of the landfilled quantities (i.e. Gobierno de España, 2008). As a result, numerous plants for C&D recovery have been opened (i.e. in Spain the plants under operation increased from 61 in 2006 to 130 in 2010, according to MMAMRM (2011)).

The refuse from C&D recovery plants constitutes a particular case of C&D waste. Several characteristics, such as the particle size or the proportion of materials making up the waste mixture, vary since recyclable fractions are separated from the original stream. Depending on the collected waste materials, organic matter (wood, paper, cardboard and others) can be a significantly great fraction among the rejects. Therefore, biodegradation processes become relevant when this kind of waste is landfilled and thus organic pollution in the leachate and gases can be problematic, like in other types of non hazardous waste landfills.

Several authors have studied the leachability of hazardous compounds from C&D waste in the last years. Roussat et al. (2008) found that one of the most troublesome components with regard to the leachate generation is wood, particularly treated and painted wood. Jambeck (2004) studied the leachability of Chromated Copper Arsenate (CCA)-treated wood as a single element and in contact with municipal solid waste (MSW) and C&D waste in lysimeters. The results regarding arsenic were presented in Khan et al. (2006), while those corresponding to the mixture with other C&D waste were compiled in Jambeck et al. (2008). Dubey et al. (2009) compared the leachate from CCA-treated wood and Alkaline Copper Quaternary (ACQ) treated wood combined with C&D waste in lysimeters. On the other hand, Wadanambi et al. (2008) analyzed the lead leachability of lead-based paint in MSW and C&D landfills environments.

Other researchers have characterized different C&D waste materials with a view to their

1 reuse. Delay et al. (2007) performed leaching tests to measure the release of inorganic  
2 pollutants in recycled C&D waste. Jang and Townsend (2001), in turn, analyzed the sulfate  
3 and dissolved solids leached from C&D debris fines. In this area, Musson et al. (2008)  
4 developed a method to approach the gypsum content in these fines and Townsend et al.  
5 (2004) studied their heavy metal leachability. Other authors have studied the influence of  
6 CCA-treated wood on the composition of mulch (Mercer and Frostick, 2012; Townsend et al.,  
7 2003).

8 Laboratory studies, like those mentioned above, are essential to gain detailed knowledge  
9 about the processes involved in this kind of landfills. However, in order for these results to  
10 contribute to the progress of technology, of the landfill engineering in this case, it is also  
11 essential to develop and publish data obtained in full scale facilities. Field data can show  
12 how experimental observations transfer to reality, where interactions among processes take  
13 place that cannot be reproduced in the laboratory. Nonetheless few data have been reported  
14 about landfills receiving C&D refuse, probably because of their recent appearance,  
15 associated with C&D recovery plants. Existing reports about C&D landfills, with or without a  
16 previous recycling facility, focus on characterizing the leachate quality or quantifying the  
17 generated sulphidric acid, as main responsible for odor nuisances, which are one of the  
18 principal social impacts of these facilities. The Environment Protection Agency of US  
19 published in 1995 a comprehensive study about the potential impacts of C&D waste landfills,  
20 including the characterization of waste and leachate in 21 landfills in the US (USEPA, 1995).  
21 Melendez (1996) compiled and compared the leachate generated in different C&D landfills.  
22 Townsend et al. (1999), in turn, studied the leachate generated by some of the materials  
23 included in this type of waste under saturated and non-saturated conditions, through several  
24 experiments in leaching columns. On the other hand, Weber et al. (2002) analyzed the  
25 leachate generated in four test cells built in a C&D debris landfill. These studies highlight the  
26 high sulfate and dissolved solid content of leachate, coming fundamentally from the  
27 breakdown of gypsum drywall (calcium and sulfate are the predominant ions observed in the  
28 leachate). Moreover, these leachates present elevated concentrations of heavy metals

1 which, in some cases such as cadmium and lead, could seriously damage human health and  
2 the environment (Melendez, 1996).

3 An important issue in C&D landfills results from the deposited gypsum. Owing to the  
4 biological reduction of sulfate from gypsum, a variety of reduced sulfur compounds (primarily  
5 H<sub>2</sub>S) is generated in C&D debris landfills, causing odor problems in many occasions, and  
6 possible health concerns (Eun et al., 2007; Lee et al., 2006; Reinhart et al., 2004). Trying to  
7 characterize the problem, Lee et al. (2006), published the gas composition of samples taken  
8 from the inside, the waste/cover interface and the ambient air in 10 C&D debris landfills. The  
9 most recent reports have focused on the search of cover materials that attenuate the  
10 emission of these compounds (Plaza et al., 2007; Sungthong and Reinhart, 2011; Xu et al.,  
11 2010).

12 These studies, the same as those about leachate mentioned above, have all been  
13 developed in American landfills. Further information is needed about the emissions of this  
14 kind of facilities in other locations, where not only the local but also the waste characteristics  
15 may be different.

16 This paper provides field data from a full-scale C&D landfill called Corral Serra in Spain  
17 (Europe). It presents the results of a five-year-study which included waste, leachate and  
18 landfill gas characterization as well as other variables measured in situ. In a first section a  
19 general description of the studied landfill and its associated C&D waste recovery plant is  
20 provided. Then, the study approach is described, including the analytical methodology and  
21 instrumentation plan. Finally, the obtained results are presented and discussed in contrast to  
22 data reported in other cases.

## 24 **2. THE STUDIED LANDFILL**

25 The new landfill of Corral Serra was built in 2005 to hold the residues from several C&D  
26 recovery plants and transfer stations in a Mediterranean region of Europe. The aim of these  
27 facilities is to generate recycled aggregate. They receive waste not only from building and  
28 civil construction works but also from demolition and building restoration. The typical

1 recovery line starts screening, separating and grinding materials with a size bigger than 400  
2 mm. Then, there is a separation of ferric metals and undesired materials materials such as  
3 plastics, metals, wood, hazardous materials, paper-cardboard, wires, glass and other  
4 materials. Finally, the sorted aggregates are crushed and divided into different sizes  
5 (normally considering three fractions: under 8 mm, under 15 mm and between 8 and 40  
6 mm). The rest (except the recovered ferric metal) are sent to a landfill.

7 The studied landfill has a total surface of 140,000 m<sup>2</sup>, of which 40,900 m<sup>2</sup> have been used  
8 for C&D waste disposal. It has a bottom lining system according to Directive 1999/31/EC  
9 (CEU, 1999). The leachate collection system consists of a 50-cm-thick layer of gravel placed  
10 on a geotextile that protects the bottom liner. A 250 mm slotted PE collection pipe receives  
11 the leachate along the South-North axis of the landfill and conveys it to the collection sump.

12 From the beginning of operations, in March 2006, until July 2009, 303,738 T of C&D  
13 materials were disposed of, including 218,697 T of waste and 85,041 T of recycled granular  
14 material used as intermediate cover. Figure 1 shows the different operation stages. During  
15 the first six months, in which 33,342 T were received, waste was spread in approximately  
16 2.5-m-high layers, without intermediate cover. During the following four months 33,142 T  
17 were disposed of, in 2-m layers with 35-cm-intermediate cover. At the end of this stage there  
18 was an average waste thickness of 5 m in Vessel 1. In the second stage 73,107 T were  
19 buried in 1.8-m-high cells with 55-cm-weekly cover during eleven months, up to an average  
20 thickness of 14 m in Vessel 1. At that time the Vessel 1 surface was at the level of the  
21 Vessel 2 bottom. Since December 2007 until July 2009 93,040 T of waste were spread in  
22 2.1-m layers with 20-cm-intermediate cover. The final average thickness was 17 m in Vessel  
23 1 and 6 m in Vessel 2.

24 Based on topographical data, a global average density (waste and cover) of 0.83 T/m<sup>3</sup> was  
25 reached in the landfill; the average apparent density of waste was 0.60 T/m<sup>3</sup>. This value is  
26 quite higher than the reference range for mixed construction waste [0.18 to 0.36 T/m<sup>3</sup>],  
27 reported by Tchobanoglous et al. (1993). This difference corresponds to the type of waste.

28 The residues received in Corral Serra come from a recovery plant that rejects the smallest

fractions, which are easier to compact than the conventional C&D waste mixture.

The yearly net precipitation in the area is negative. Taking advantage of this fact, leachate is stored inside the landfill and in external ponds, from where it is recirculated to promote evaporation by irrigation on the waste when there is no rain.

### **3. EXPERIMENTAL STUDY**

#### **3.1 WORK PLAN**

In order to take advantage of the new landfill as a pilot experiment to characterize the emissions in this kind of facilities located in similar climatic areas in Europe, a study was planned. It included a detailed record of the operational practices, waste, leachate and gas samples characterization, in situ monitoring of several variables (temperature, leachate head on the bottom liner and settlement) and gas/liquid samples extraction from several points inside the waste body. This study was as a part of a Collaboration Research Project between TIRME S.A. and the Environmental Engineering Group of the University of Cantabria (GIA-UC), to gain knowledge about processes occurring inside the landfill and help understand the consequences of leachate recirculation, which is not a common practice in this kind of landfills. It supplements the general Environmental Surveying Plan of the landfill, which also considers surface gas analyses by flux chamber method monthly, quarterly analyses of underground water, leachate and runoff water and topographical surveys.

#### **3.2 INSTRUMENTATION**

An instrumentation plan was designed for monitoring some internal parameters in the landfill (temperature, leachate head on the liner, gas composition and settlement). Considering the control needs, financial constraints and the operation plan, the landfill was divided into several zones, where instrumentation was installed. During the operation period four PT-100 thermoresistences (T), four vibrating wire piezometers model 4500 (Geokon) (PZ), two of them with temperature reading, six gas/liquid sampling ports (SP) and 2 settlement plates were installed. The sampling ports are handmade devices based on Zhao et al. (2003), consisting of a slotted stretch of PVC pipe connected to the exterior through 15-mm PVC

1 hoses, from which gas or liquid samples, depending on what is accumulated in the  
2 corresponding point, are pumped. In Vessel 1, PZ 1-1 and PZ 2-1 were placed at an  
3 elevation of 36.3 m, T1-2 and T2-2 at 42.5 m and T1-3 and T2-3 at 50 m. Besides, four  
4 sampling ports were installed in the same landfill vessel (SP1-2 and SP2-2 at a height of 6.5  
5 m and SP 1-3 and SP 2-3 at 14 m from the bottom) and two in Vessel 2 (SP3-1 and SP4-1 at  
6 0.5 m from the bottom) (Figure 1).

7 Piezometers and thermometers are manually read (via GEOKON GK-404 model and ISO-  
8 TECH IDM 91E digital multimeter, respectively) every week. On the other hand, a portable  
9 gas measurement device (GA 2000, Geotechnical Instruments) is used for the analyses of  
10 gas samples taken from the sampling ports installed within the waste. The detection limits of  
11 the equipment are 0-100% for CH<sub>4</sub> and CO<sub>2</sub>, 0-25% for O<sub>2</sub> and 0-500 ppm for H<sub>2</sub>S and CO.  
12 Every six months a vacuum pump (Millipore XX5522050) is connected to the hoses to obtain  
13 the samples of the gas/liquid in contact with the waste. After a first suction period of 10 min,  
14 if no liquid is obtained, the pump exit is connected to the GA2000 meter, with which the gas  
15 is suctioned for 60 s and analyzed. The measurement with the portable analyzer is repeated  
16 for 30 s. Then the pumping is started for five minutes and connected to the GA2000 again.  
17 The four measurements obtained as a result are compared and, if they are stable, averaged  
18 to give the final value. Otherwise the suction time is increased, until a steady reading is  
19 obtained.

20 Two settlement plates were placed on Vessel 1, but they had to be eventually removed due  
21 to operational interferences in the active area.

### 22 **3.3 PERFORMED ANALYSES**

23 Five characterization campaigns were performed since the landfill opening to determine the  
24 material composition of the waste incoming from each origin (a total of 8 plants). The  
25 campaigns took place in June 2007, November 2007, June 2008, November 2008 and June  
26 2009.

27 From June 2007, leachate samples taken from the collecting sump and gas collected from  
28 the sampling devices inside the waste have been analyzed with different frequencies in each



1 operation stage. Leachate analyses were initially performed on a monthly basis, every two  
2 months later and half-yearly after the end of operation in July 2009. They include pH,  
3 dissolved oxygen (DO), conductivity, redox potential (ORP), dissolved and total chemical  
4 oxygen demand (COD), dissolved and total biochemical oxygen demand (BOD<sub>5</sub>), total  
5 organic carbon (TOC), total nitrogen (TN), ammonia nitrogen (NH<sub>4</sub>-N), alkalinity, sulphates,  
6 total solids (TS), total volatile solids (TVS), total suspended solids (TSS) and volatile  
7 suspended solids (VSS), and heavy metal concentrations. All the analyses are performed  
8 according to standardized methods (Eaton et al., 2005). For metal concentrations, in  
9 particular, leachate samples were preserved with concentrated acid nitric (HNO<sub>3</sub>) (pH<2).  
10 Then, they were digested according to the Standard Method 3030E and filtered afterwards  
11 with a 0.45 µm filter in order to be analyzed in an atomic absorption spectrometer (Perkin  
12 Elmer 300 AAnalyst Spectrometer).

13

## 14 **4. RESULTS**

### 15 **4.1 WASTE CHARACTERIZATION**

16 The waste received in the landfill was recorded distinguishing the quantities coming from  
17 different sources. Considering that record and also the results of the different  
18 characterization campaigns, the composition of the waste disposed in Corral Serra was  
19 obtained (Table 1).

20 As it is shown, slowly (wood, textiles) and readily (paper and cardboard) biodegradable  
21 materials represent a great fraction of the refuse stream. The significant contribution of  
22 wood, in particular, 31% of the total disposed material, coincides with the amount received in  
23 other C&D landfills (USEPA, 1995). In fact, great part of the experimental studies on the  
24 pollution potential of C&D waste, some of them mentioned above, focus on characterizing  
25 the leachability of wood treated in different ways (Dubey et al., 2009; Jambeck et al., 2008;  
26 Khan et al., 2006). Wood, in this case, comes mainly from furniture collected in the transfer  
27 stations and demolition of buildings where it is used in some elements (windows and  
28 shutters, doors, floors and, rarely, in structures). Its high presence in the landfill is due to the

1 fact that nowadays it is not recovered in recovery facilities.

2 Apart from that, the small quantity of gypsum identified in this case (0.1% w/w) is

3 remarkable. It coincides with the range reported for this kind of waste in Europe, between 0.1

4 and 0.4% (EU, 2011). However, the actual fraction of this material may be greater, if we

5 consider the gypsum contained in the fines. According to the operator company the average

6 fraction of gypsum in the fines recovered in the Material Recovery Facilities (MRFs) in the

7 area is 3.6%. On the other hand, Musson et al. (2008) the gypsum represents between 1

8 and 25% of the fines generated in different industries in the US. Assuming that the fines in

9 this study included 25% of gypsum, the percentage of this material in the landfill studied in

10 this case would be 3.7%, still very limited compared to data stated in other countries.

11 USEPA (1998) reports a range from 21 to 27% of gypsum drywall in the C&D waste received

12 in American landfills. This great difference is caused by the specific construction customs in

13 each region. Bricks are usually used to partition off buildings in the South of Europe and not

14 gypsum drywall. The only gypsum building elements commonly used in those countries are

15 plaster ceilings and plaster mouldings.

16

## 17 **4.2 LEACHATE VOLUME AND COMPOSITION**

### 18 **4.2.1 Leachate volume**

19 During the 66 months of study, 1221 mm (an average of 18 mm/month) of net precipitation

20 (direct precipitation on the landfill surface minus potential evaporation, on a monthly balance)

21 were received. Figure 2 shows the rainfall intensity in the landfill and the collected leachate

22 during the months studied. As it is expected in a small landfill with important preferential

23 pathways, the response to rain is fast. The delay between rainfall and collected leachate that

24 appears in the graphic is mainly due to the way in which it was managed, with temporary

25 water storage inside the landfill. During the study period, leachate was being collected from

26 the landfill discontinuously, to recirculate it directly on the waste or to store it in a regulation

27 pool (for subsequent recirculation).

1 Until August 2011, 393 mm of leachate (6 mm/month on average) were collected (a total of  
2 16,057 m<sup>3</sup>). It means 13% of the rainfall registered throughout the time (2,925 mm), that is, a  
3 very low value especially taking into account the leachate recirculation. As a reference,  
4 Stegmann and Ehrig (1989) estimated an annual average ratio of leachate/rainfall of 18% in  
5 a study of several landfills in Germany, operating without leachate recirculation.

6 Bearing in mind the quantity of waste disposed during the operation, it leads to a global  
7 average L/S ratio of 53 L/T during those months (16,057 m<sup>3</sup> over 303,738 T), quite low when  
8 compared to other values reported in studies of C&D waste. For example, Weber et al.  
9 (2002) added between 2,800 and 4,400 L of rainfall per T of waste during six months of  
10 experiment to characterize the leachate from four test cells filled with residential construction  
11 waste. On the other hand, Delay et al. (2007) achieved a L/S ratio of about 600 L/T after 400  
12 days in their column leaching test to determine the quantity of inorganic contaminants  
13 released in that kind of waste.

#### 14 **4.2.2 Leachate composition**

15 Table 2 shows the maximum, minimum and average values obtained in the leachate  
16 analyses during this period compared to other data reported in the literature for C&D landfill  
17 leachate. The average values observed are generally higher than the reference average  
18 values compiled by Melendez (1996), as befits a dry landfill like Corral Serra. A classic  
19 leaching curve for inorganic elements such as those presented by Delay et al. (2006) for  
20 demolition waste, shows a front where leachate concentrations remain in the maximum  
21 values, given by the solubility of each element, during the first stages of the experiment.  
22 After a minimum L/S ratio (above 0.1 L/kg, depending on the element studied)  
23 concentrations begin to decrease as more water is added, reflecting the washing-off effect.  
24 The ratio achieved in this study (0.053 L/kg) is well under the typical minimum, what shows  
25 that high concentrations observed correspond to the front of the leaching curve and will  
26 decrease in future as water washes the waste.

27 Besides, the leachate recirculation carried out since June 2007 favors evaporation of water  
28 and therefore the reduction of the volume to manage and, consequently, the increase of its

1 concentration. Furthermore, recirculation raises the moisture content of waste and improves  
2 the distribution of water and its nutrients. This facilitates waste degradation, both by physico-  
3 chemical and biological phenomena, and thus the solution of pollutants into the leachate. As  
4 a result, in this case some samples exceed the maximum reference concentrations in some  
5 parameters: conductivity, ammonia nitrogen, arsenic and lead.

6 As in other places, the pH values maintained in the neutral-basic range (between 7.4 and  
7 8.3). This matches the high alkalinity (an average of approximately 3,200 mg CaCO<sub>3</sub>/L) due  
8 mostly to dissolution of carbonates from concrete and disposed aggregates. The average  
9 conductivity registered in this study (8.3 mS/cm) is five times higher than the average found  
10 in bibliography. It matches the total dissolved solids (TDS) concentrations found in this  
11 leachate which, although they do not go beyond the reference range (990 – 8,400 mg/L in  
12 Melendez (1996)), their average (4,860 mg/L) is twice the average values reported.

13 Conductivity and TDS are related to the quantity of dissolved ions. In this leachate the  
14 largest ions are alkalinity, sodium, calcium and sulfates. The first three of them appear in  
15 higher concentrations than those found in literature. The high concentrations of sulfates are  
16 specially striking in this case, being the quantity of gypsum smaller than in other studies;  
17 they are a consequence of the limited L/s ratio mentioned above together with the high and  
18 fast solubility of gypsum (see Musson et al., 2008).

19 Conductivity, sodium and sulfates measured in the leachate samples are shown in Figure  
20 3A. The temporary effect of dilution caused by rainfall, which reduces leachate concentration  
21 in October 2007, May and December 2008 and October 2009, for instance, is seen. Taking  
22 into consideration the great variability depending on rainfall, an increasing trend throughout  
23 the study period can also be noticed. That slight trend, an effect of the leachate  
24 concentration by recirculation, is also observed in sulfate and sodium concentrations.

25 Figure 3B shows the COD and NH<sub>4</sub>\_N evolution over time. In comparison with other  
26 parameters, these ones do not indicate a clear trend; they keep values in a similar range for  
27 the whole period. In the case of COD, its trend to increase may be made up for by the  
28 transformation of biodegradable compounds into gas. These phenomena are discussed in

1 the paragraphs below.

2 COD concentrations within the published range have been registered but, once again, with  
3 higher average concentrations. Among C&D waste, the materials which cause the most  
4 organic pollution are: cardboard, insulation materials and wood (Townsend et al., 2000). In  
5 the case studied, these components represent more than 36% of the disposed material.  
6 However, because of the type of waste, the organic pollution is essentially non-  
7 biodegradable: the BOD/COD relation has fluctuated between 0.044 and 0.205.

8 Compared to other authors,  $\text{NH}_4\text{-N}$  average concentration (400 mg/L) stands out, thus they  
9 are significantly higher (more than 10 times) than the average reported value. Since the  
10 source of  $\text{NH}_4\text{-N}$  is the decomposition of organic waste, these high concentrations can be  
11 associated to the amounts of wood and cardboard disposed of, to the enhanced  
12 decomposition of the waste and, above all, to the limited dilution of the leachate in this case.

13 On the other hand, part of the  $\text{NH}_4\text{-N}$  which appears in the leachate volatilizes in the form of  
14 free ammonia. The rate of free ammonia in the leachate depends on pH and temperature.  
15 Ritzkowski and Stegmann (2003) found in an aerobic bioreactor that with a pH 7.4 and at  
16 35°C, the 50% of ammonia nitrogen initially present in the leachate was volatilized. Thus, the  
17 quantity of  $\text{NH}_4\text{-N}$  volatilized can be important in the conditions of the landfill studied  
18 (moderately basic pH and average temperatures between 30 and 51°C), especially in the  
19 recirculation by superficial spraying. In fact, a smell of ammonia is perceived during leachate  
20 recirculation around the recirculating truck. Moreover, when waste samples have been  
21 extracted by means of drilling or excavation that smell has also been detected, sign of the  
22 volatilization of the compound in the environment. This explains why its concentration in  
23 leachate has not increased during these years of operation.

24 On the other hand, as it is usual in this kind of landfills, several heavy metals were  
25 consistently detected in the leachate. Due to the methodology used, the measured  
26 concentrations include the particulate fraction of metals present in the samples, which were  
27 digested before the filtration. Because of that some values exceed the solubility  
28 corresponding to the pH detected, as it happens in other studies. Pb and As stand out

1 among them. Both the range and the average of arsenic concentrations obtained in the  
2 leachate exceed the typical values found in C&D landfills. Chromium is within the range, but  
3 with higher average concentrations, while copper is in a lower range. These three elements,  
4 in various chemicals forms, are used to preserve wood (Melendez, 1996). It is the case of  
5 CCA treated wood, preserved with chromic acid, copper oxide and arsenic acid (USDA,  
6 1980) to which Weber et al. (2002) attribute most of As, Cu and Cr found in landfill. As well  
7 as the case of ACQ-treated wood, preserved with alkaline copper quaternary, whose main  
8 ingredient is copper (Dubey et al., 2009). Higher arsenic concentrations, than those obtained  
9 in this study (48 - 724  $\mu\text{g As/L}$ ), were found in specific studies about leachability of treated  
10 wood. Dubey et al. (2009) used 5% by weight of CCA-treated wood in the analyzed waste  
11 and measured As concentrations from 500 to 1,200  $\mu\text{g/L}$ , while Jambeck et al. (2008) used  
12 10.2% CCA-treated wood and measured concentrations ranging from 1,090 to 4,250  $\mu\text{g/L}$   
13 (with a L/S range of 1560 L/T, quite higher than the present study). However, Jang and  
14 Townsend (2003) (0.5% CCA-treated wood) measured lower concentrations (10 - 380  $\mu\text{g/L}$ ).  
15 On the other hand, Cr concentrations measured in this leachate range from 5 to 250  $\mu\text{g/L}$ .  
16 These values are similar to those obtained by Jang and Townsend (2003) (71.9 - 165  $\mu\text{g/L}$ )  
17 and by Dubey et al. (2009) (75 to 200  $\mu\text{g/L}$ ), but lower than the ones obtained by Jambeck et  
18 al. (2008) (300 to 2,100  $\mu\text{g/L}$ ).

19 In this case, with regard to the studies of reference, the lower presence of Cu may be due to  
20 its combination with reduced sulfur compounds forming very insoluble precipitates which  
21 remain in the waste within the landfill (Dubey et al., 2009; Jambeck et al., 2008; Weber et al.,  
22 2002). This combination and other ones of different elements are promoted by recirculation,  
23 which increases the retention time of these compounds and improves their mixture in the  
24 whole landfill.

25 In the case of lead, both the range and the average obtained in this study have been higher  
26 than concentrations found in literature, although only on six occasions (of 31 measurements)  
27 have exceeded 1,500  $\mu\text{g/L}$ . The main sources of lead in C&D waste are paints and coating  
28 used in the past to cover wood surfaces and other materials, caulking products in which lead

1 is an additive and is also used in flashing (Melendez, 1996).

2 Figure 3C shows the evolution of As, Cr, Cu and Pb concentrations in the leachate. As it is  
3 presented, As, Cu and Pb concentrations have decreased during these months. This drop  
4 may be connected with the evolution of the redox potential (ORP) which has changed from  
5 values lower than -300 mV in the second year of monitoring to values higher than 350 mV at  
6 the end of the study. In the case of As, for example, its mobility drops at a constant pH with  
7 higher ORP values and as a result, the amount of leached arsenic decreases (Moghaddam  
8 and Mulligan, 2008).

## 10 **4.3 INSTRUMENTATION**

### 11 **4.3.1 Gas composition**

12 No odor nuisances, typical of these C&D landfills (i.e. Lee et al. (2006) reported ambient  
13 concentrations up to 50 ppmv), have been reported by the closest residents and landfill  
14 workers in Corral Serra. In fact, considerable concentrations of landfill gases were not  
15 detected in the air quality monitoring campaigns within the landfill boundaries. Likewise, no  
16 significant surface emissions were measured through the dynamic flux chamber method with  
17 latter gas chromatography (GC) analyses (CEN EN 13725, 2003): CH<sub>4</sub> and H<sub>2</sub>S remained  
18 under detection limits, of 10 ppm and 0.001 ppmv respectively, and CO<sub>2</sub> concentrations  
19 between 0.04 and 0.14% were measured.

20 However, analyses of gas collected from the sampling ports installed inside the waste  
21 allowed detecting a certain anaerobic activity in the landfill, in keeping with the leachate  
22 characteristics and its high retention time, which favors decomposition of dissolved  
23 pollutants. Figure 4 presents the evolution of CH<sub>4</sub> and H<sub>2</sub>S concentrations registered in the  
24 sampling ports 1-2, 1-3 and 4-1 over time: in order not to make the graph difficult to interpret  
25 these points have been chosen as representative of the different layers where  
26 instrumentation inside the landfill has been placed. As it is shown, maximum concentrations  
27 of 50% of CH<sub>4</sub> and above 500 ppmv of H<sub>2</sub>S were reached in these internal points of the  
28 landfill. Following methane concentrations, how the degradation processes developed in

1 several areas of the landfill can be estimated.

2 Methane concentration increased quickly in the intermediate layers of Vessel 1 (represented  
3 by the point SP 1-2), since the installation of the sampling ports. In March 2008 this  
4 concentration started decreasing. At that moment there were already more than 10 m of  
5 waste over the sampling ports which made the flow of water through the area difficult. This,  
6 along with the existing high temperature (around 50°C according to the monitoring data),  
7 dried the waste in the area.

8 In contrast, in Vessel 2 (point 4-1) the methane rises more slightly as the waste is covered.  
9 This SP is buried near the bottom of the vessel and thus it does not receive contributions of  
10 gas coming from older waste located beneath it, such as the case of SP1-2, mentioned  
11 above.

12 The latter sampling ports placed (SP 1-3) stayed two meters below the final level reached in  
13 area 1, leaving 14 m of waste below. They quickly detected significant concentrations of  
14 landfill gas; at first, most of it came from the lower layers and then, they were almost  
15 certainly generated in the area, which remained moist thanks to the surface water  
16 contribution.

17 These results were obtained sucking previously the hoses for ten minutes, as it has been  
18 explained in the section on methodology. On the whole, stable measurements were not  
19 achieved when the length of suction was shorter. This means that the pressure of biogas  
20 (and therefore, its generation) on the inside is limited. This is also revealed in the  
21 measurements registered in January 2010, which correspond to a period of great intensity of  
22 leachate recirculation which caused a marked drop in temperatures (Figure 5) and displaced  
23 the gas generated in the waste pores, reducing sharply the measured concentrations. This  
24 limited pressure, together with the use of intermediate cover layers during the operation,  
25 explains the lack of significant emissions through the surface.

26 As for hydrogen sulphide, it shows a different trend, common in all the points. The highest  
27 concentrations were observed in the beginning owing to sulfate degradation (previous to the  
28 implementation of the methanogenic phase). In other similar landfills soil vapor



1 concentrations have been measured up to 12,000 ppmv (Lee et al., 2006). However, in this  
2 case the smaller drywall presence in waste makes  $\text{SO}_4^{2-}$  availability more limited and thus  
3 the production of  $\text{H}_2\text{S}$  by sulfate reducing bacteria is lesser. Besides, there are other  
4 phenomena that reduce  $\text{H}_2\text{S}$  concentrations in air, such as cover soil attenuation, air dilution,  
5 its sorption to concrete surfaces (Plaza et al., 2007) or its solution into the leachate (Reinhart  
6 et al., 2004). In any case, during the first stages of operation, in 2006, there was a significant  
7 concentration of hydrogen sulphide in leachate: yellow precipitates were formed in the  
8 leachate collection sump, which probably originated from spontaneous sulphide oxidation to  
9 sulfur when the leachate came out the landfill to an oxic ambient. This is effect of the little  
10 dilution that causes, despite the limited presence of sulphated elements in the waste,  
11 leachate to show high concentrations of these compounds.

#### 12 **4.3.2 Temperatures**

13 Figure 5 shows the evolution of the temperature lectures in three of the installed PT100, in  
14 contrast to the ambient temperature in the landfill. As in the case of sampling ports, these  
15 sensors have been chosen as representative of each landfill area in order to simplify the  
16 graph and make its interpretation easier.

17 Thermocouples 1-2 and 2-2 were the first ones placed inside the waste, 6.5 m from the  
18 bottom of Vessel 1. During the operation 7.5 m of waste were placed over them. A rapid rise  
19 in the temperature, from over 30°C up to the thermophilic range (50-60°C), took place during  
20 the first months after the sensors installation. These high temperatures indicate a release of  
21 thermal energy as a result of waste degradation and coincide with the maximum data  
22 obtained in other landfills (i.e. Yeşiller et al. (2005)). After that, temperatures started  
23 decreasing until they reached values around 42°C at the end of the study, that is, in the  
24 appropriate range for methane generation (Rees, 1980).

25 Thermocouple 1-3, placed in the same area but 14 m from the bottom, registered a similar  
26 initial increase (from 49 to 57°C in the first months) which dropped immediately when it  
27 started to rain in the fall of 2009. From that moment it developed in parallel with the ambient  
28 temperature. It is important to take into account that this sensor, along with the 2-3 one, had

1 less than 2 m of waste above, unlike 1-2 and 2-2.

2 The evolution of temperatures at point 4-1, in turn, is much smoother. It was located on the  
3 bottom of Vessel 2, isolated from the outside by an up to 5.5-m-thick layer of waste. It is also  
4 in contact with the water which percolates through the waste and goes to the leachate  
5 extraction pipe. An increasing trend in time is observed, which matches an active waste  
6 degradation.

### 7 **4.3.3 Piezometers**

8 During the 41 months water storage was constantly registered only in PZ 1-1, which is  
9 located at the lowest level (36.3 m) of the four piezometers of control. In this point readings  
10 began increasing continuously from July 2009, in keeping with the strategies of leachate  
11 storage within the waste. Since then, they stayed on average values of 36.9 m, revealing  
12 water storage on the bottom of approximately 60 cm. As a consequence of rainfall, isolated  
13 positive readings (some weeks) were registered both at this point and at the other  
14 piezometers during operation. These monitoring data of water storage at the bottom were  
15 used to control the strategy of leachate management and to check the correct operation of  
16 the drainage system.

## 18 **5. CONCLUSIONS**

19 In spite of its origin, which is a specific facility for C&D waste recovery, the materials  
20 disposed in the studied case are similar to other C&D debris sites, mainly made up of wood  
21 and concrete. Nonetheless, the amount of gypsum drywall is significantly smaller than the  
22 reported in the available American studies. This makes the behavior of the landfill different,  
23 as it will be in other sites, depending on the composition of the waste. A significant difference  
24 is that no odor nuisances are expected in these cases, even though landfill gas can be  
25 generated through biological degradation of organic waste.

26 The amount of water received in the landfill is significantly smaller than in other cases  
27 reported, due to the scant precipitation and large evaporation potential in the area. As a  
28 result, the generated leachate volume is considerably limited. This supports the strategy

adopted for the leachate management in cases like this one, that is, with internal storage in the landfill or external one in the pond and recirculation when it does not rain.

In regard to the liquid emissions, an outstanding presence of dissolved salts and metals, as observed in other cases, has been measured. The recirculation and the great evaporation make the leachate to be quite concentrated. Because of that, some parameters in this study exceed the ranges reported in the literature, showing the key influence of the liquid to solid ratio to which waste has been subject. Such is the case of conductivity, ammonia nitrogen, lead and arsenic. It must also be noticed that recirculation facilitates waste degradation when its moisture increases although in this case in a very limited way. This effect also supports the approach adopted for a more sustainable operation.

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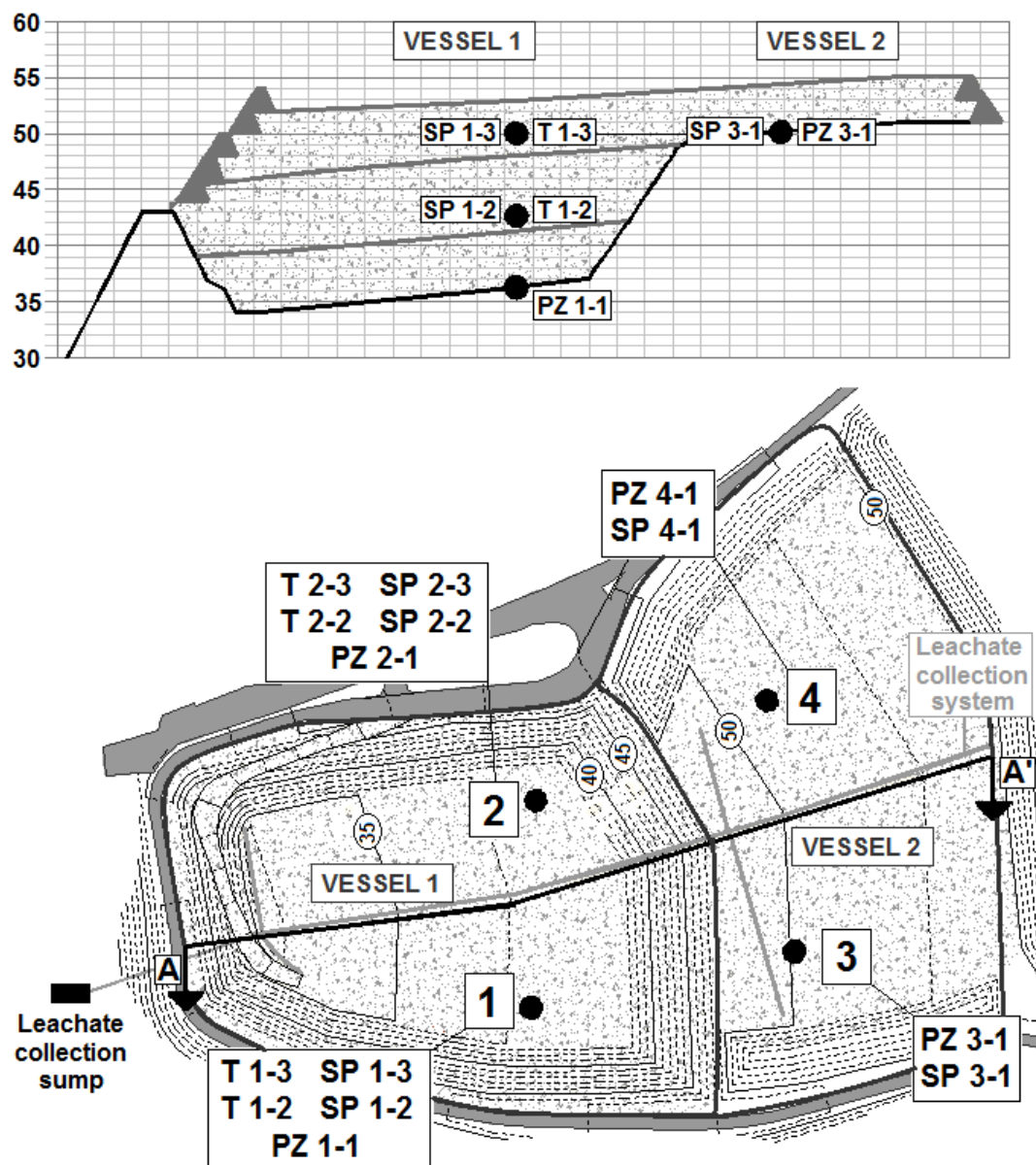
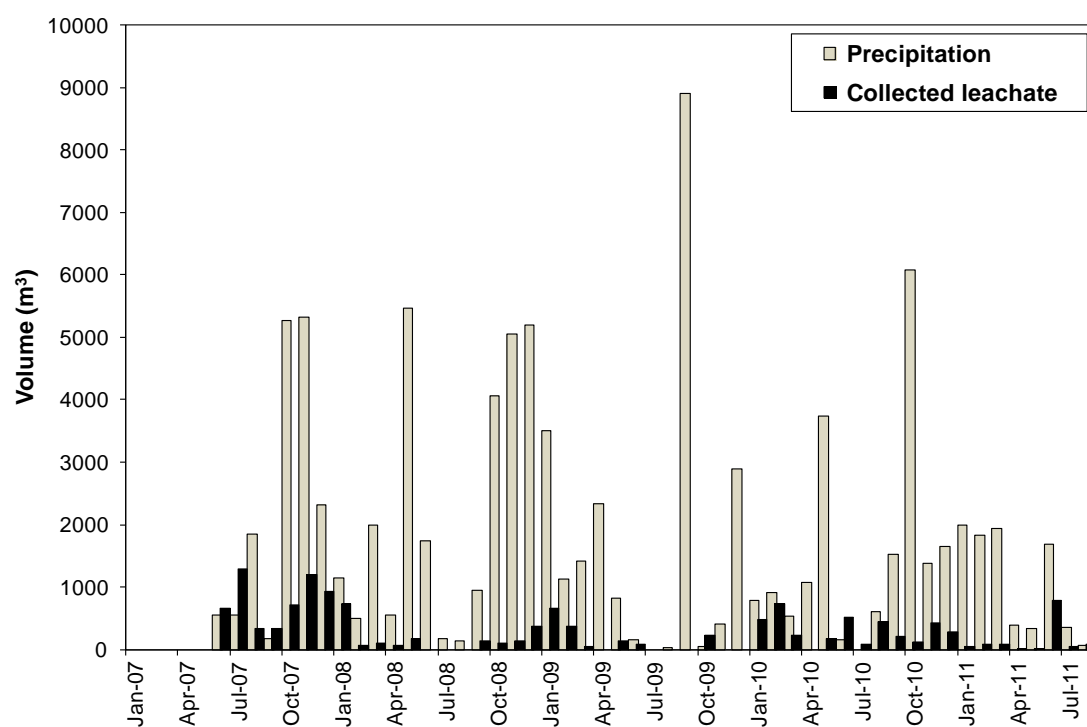


Figure 1. Studied landfill: layout, operation phases and instrumentation





**Figure 2. Precipitation and collected leachate during the studied period**

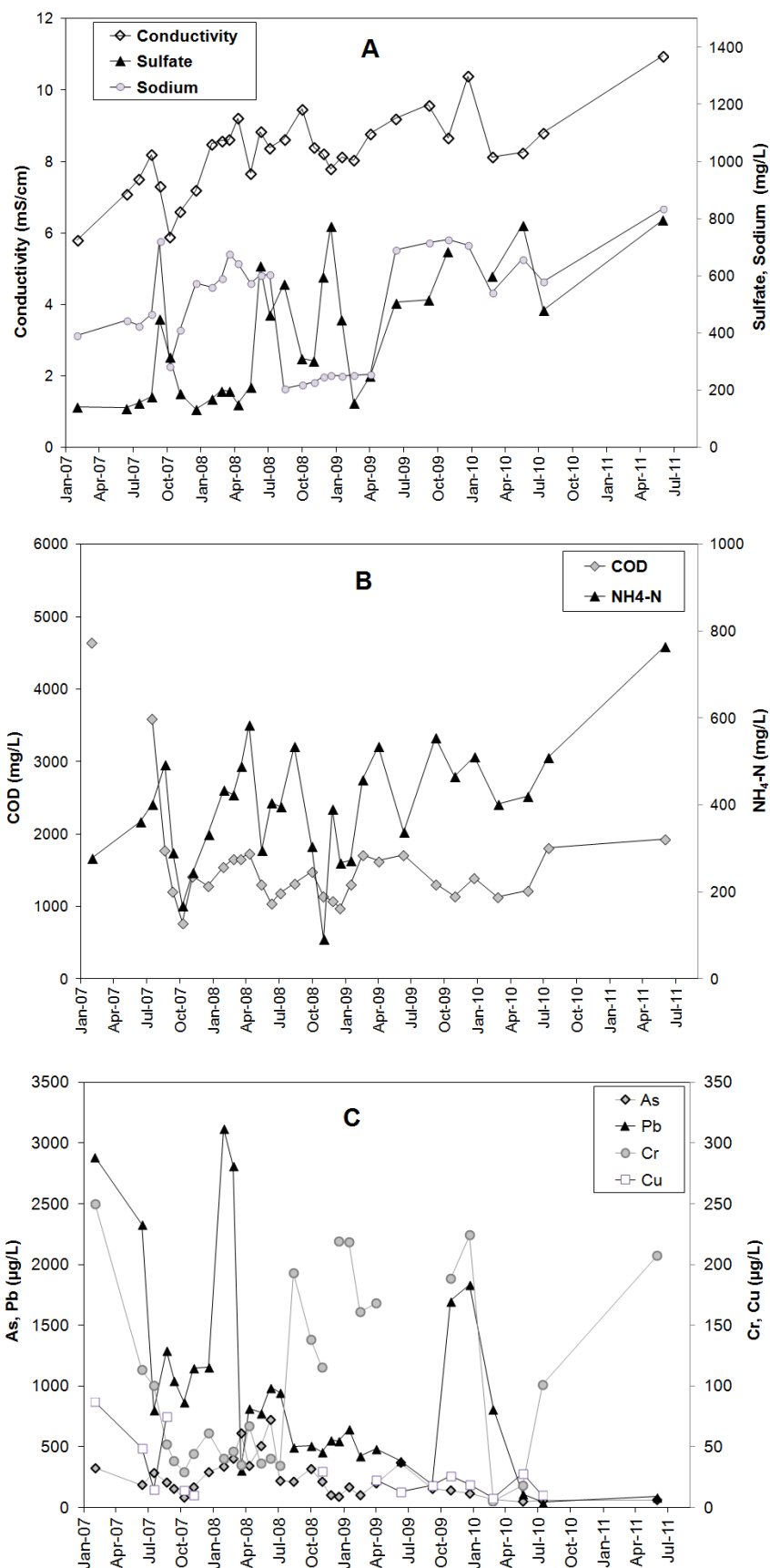
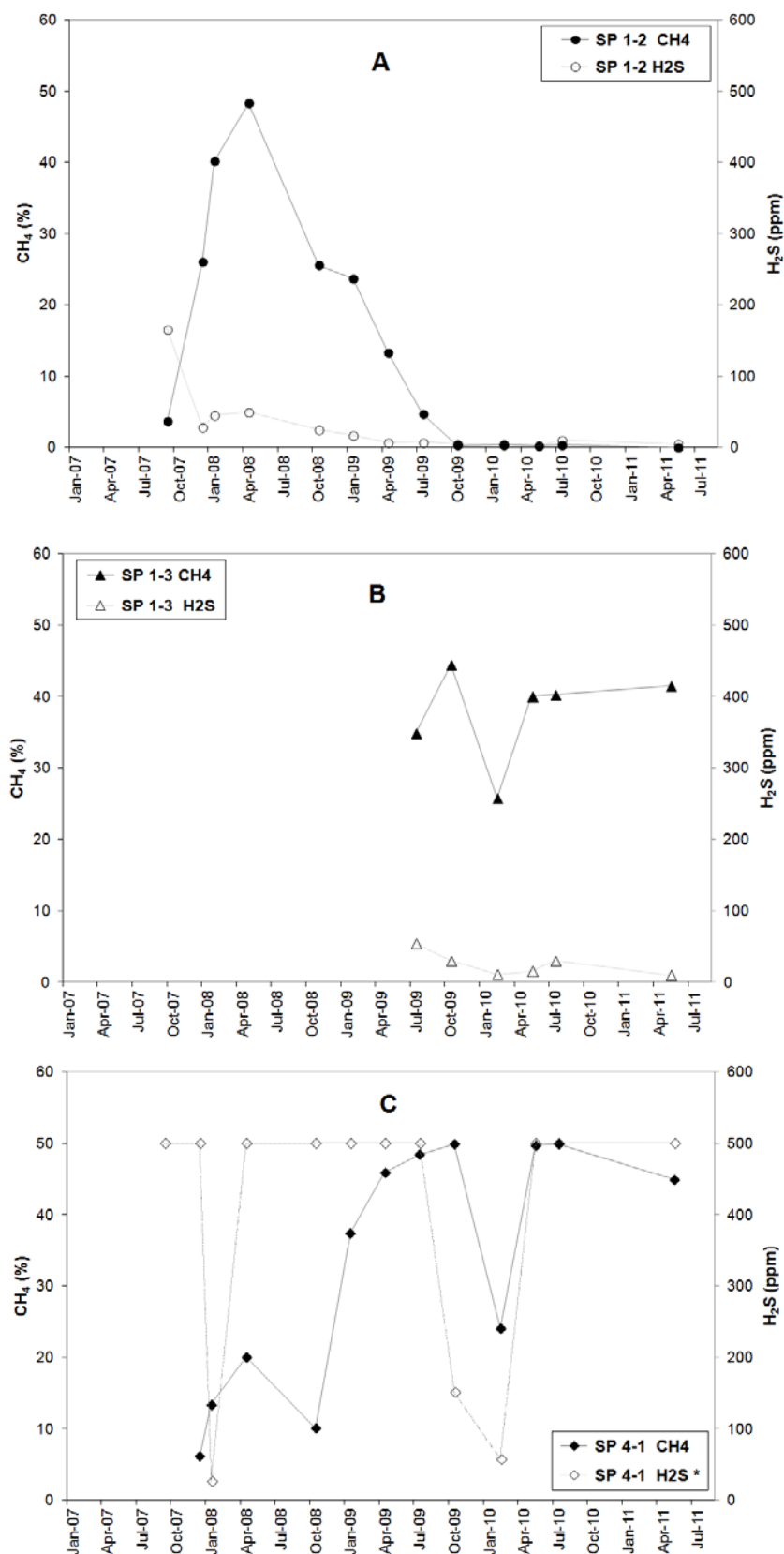
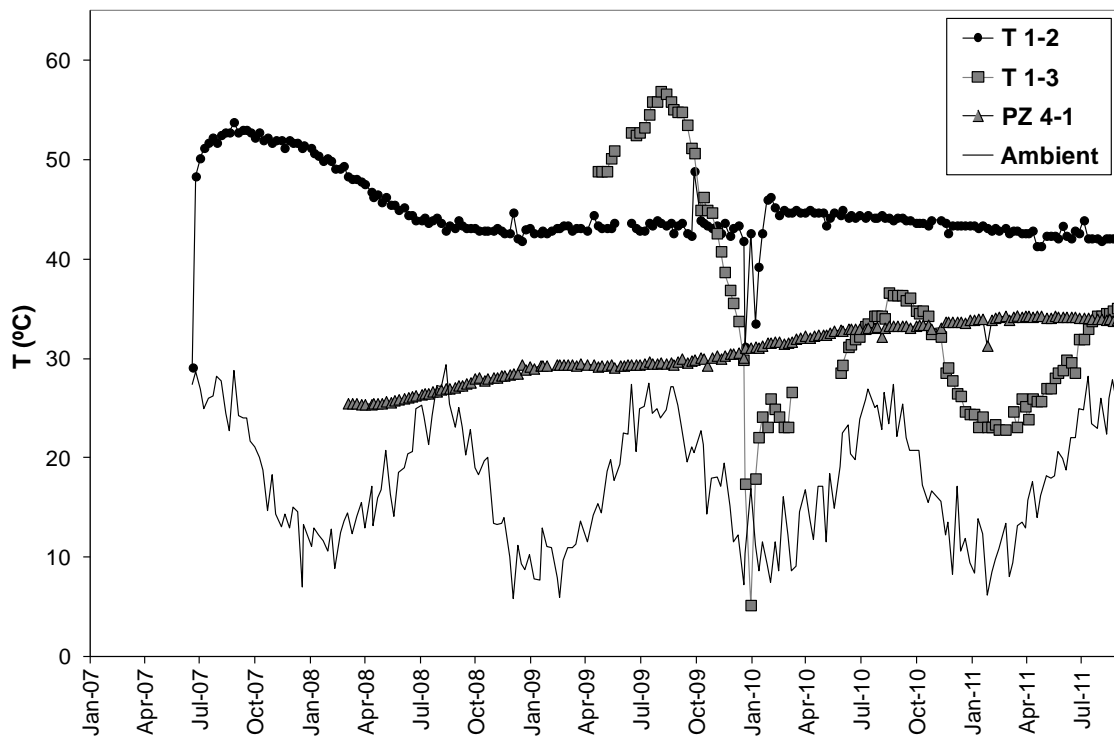


Figure 3. Pollutant concentrations in the leachate



\* values presented as 500 ppm correspond to out of range readings (GA2000 range: 0-500 ppm)

**Figure 4. CH<sub>4</sub> and H<sub>2</sub>S in the sampling ports inside the landfill**



**Figure 5. Temperatures inside the landfill during the studied period**

**Table 1. Material dumped in the studied landfill.**

<b>Material</b>	<b>Total weight (T)</b>	<b>% by weight</b>
Aggregates	85,041	28.0
Inert building materials (bricks, concrete, stones)	15,309	5.0
Fine inert fraction	44,177	14.5
Ferric metal	6,124	2.0
Non ferric metal	1,093	0.4
Glass	1,531	0.5
Gypsum	219	0.1
Ceramics o porcelain	437	0.1
Cardboard	13,559	4.5
Plastic	20,339	6.7
Rubber	1,968	0.6
Wood	95,571	31.5
Textile	13,559	4.5
Wire	1,312	0.4
Expanded polystyrene and foam	3,499	1.2
<b>TOTAL</b>	<b>303,738</b>	<b>100.0</b>

**Table 2. Leachate quality results versus published literature values**

Parameter	Units	Studied landfill <sup>a*</sup>	USEPA, 1995 <sup>b</sup>	Melendez, 1996 <sup>c*</sup>	Townsend et al., 2000 and Weber et al., 2002 <sup>d*</sup>	Wang et al., 2012 <sup>e*</sup>
pH		6.8-8.3 (7.5)	6.2 - 8	4.45 – 8 (6.95)	6.1 – 7.9 (6.90)	6.1 – 6.9 (6.4)
DO	mg/L	0,3-2.1 (1.0)			0.06 – 1.58 (0.5)	
Conductivity	mS/cm	5.8-11 (8.3)		(1.67)	1.1 – 3.1	
ORP	mV	-407/392 (-89)			< -200	
Total COD	mg/L	775-4,641 (1,571)	11,200	11,200 (755)	115 - 700	
Dissolved COD	mg/L	586-4,190 (1,407)				
Total BOD <sub>5</sub>	mg/L	70-500 (227)	320	920 (87)		
Dissolved BOD <sub>5</sub>	mg/L	20-150 (99)				
Dissolved TOC	mg/L	120-1,185 (404)	1,080	2,100 (307)		
Alkalinity	mg CaCO <sub>3</sub> /L	1,800-4,170 (3,189)	6,520	938.2 – 6,520 (965)	210 – 960 (530)	75 – 725
NH <sub>4</sub> -N	mg/L	92-765 (401)	305	305 (13)	<1 – 4.1	
Dissolved TN	mg/L	182-844 (463)				
Sulfates	mg/L	133-1,038 (405)	2,700	11.7 – 1,700 (254)	310 – 1,370 (880)	
TS	mg/L	3,756-5,776 (4,939)				
TDS	mg/L	3,412-5,765 (4,860)	8,400	990 – 8,400 (2,263)	970 – 3,310 (2,120)	873 – 2,010
TVS	mg/L	1,208-2,472 (1,619)	170 - 380			
VSS	mg/L	5-781 (75)	43,000			
Calcium	mg/L	28-608 (150)	600	90 – 600 (270)	225 – 690 (470)	274
Sodium	mg/L	206-834 (495)	1,510	11 – 1,290 (163)	18.8 – 100.3 (42.8)	21 – 37
Chromium	µg/L	5-250 (105)	250	250	6 – 74.9 (17.8)	
Cadmium	µg/L	< 2-182 (27)	2,050	2,050 (31.9)	ND	
Copper	µg/L	< 1-87 (28)	620	5 – 620 (20.3)	5.6 – 1,740 (92)	
Zinc	µg/L	21-735 (276)	8,630	8,630 (657)	< 100 – 1,731 (433)	
Lead	µg/L	43-3,119 (987)	2,130	4.9 – 2,130 (8.8)	< 1 – 14.1 (4.1)	
Nickel	µg/L	< 3-152 (59)	170	30 – 170 (20)	ND	
Arsenic	µg/L	48-724 (233)	120	1.4 – 77.3 (12.3)	< 10 – 148 (43.8)	<4
Mercury	µg/L	< 2-4.3 (1.4)	9	9	ND	

a Concentration ranges in the studied landfill.

b Full-scale study.

c Full-scale study. Concentration ranges from literature review of C&D leachate.

d Field-cell average

e Lab-scale C&D lysimeter study.

\* Values in parentheses indicate the average value found for each parameter

