

Structural Correlations in Jahn-Teller Systems of Mn^{3+} and Cu^{2+} : Unraveling Local Structures through Spectroscopic Techniques

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1. Electronic d -orbital structure of Cu^{2+} in $\text{KZnF}_3:\text{Cu}^{2+}$ and $\text{K}_2\text{ZnF}_4:\text{Cu}^{2+}$ from optical spectroscopy.

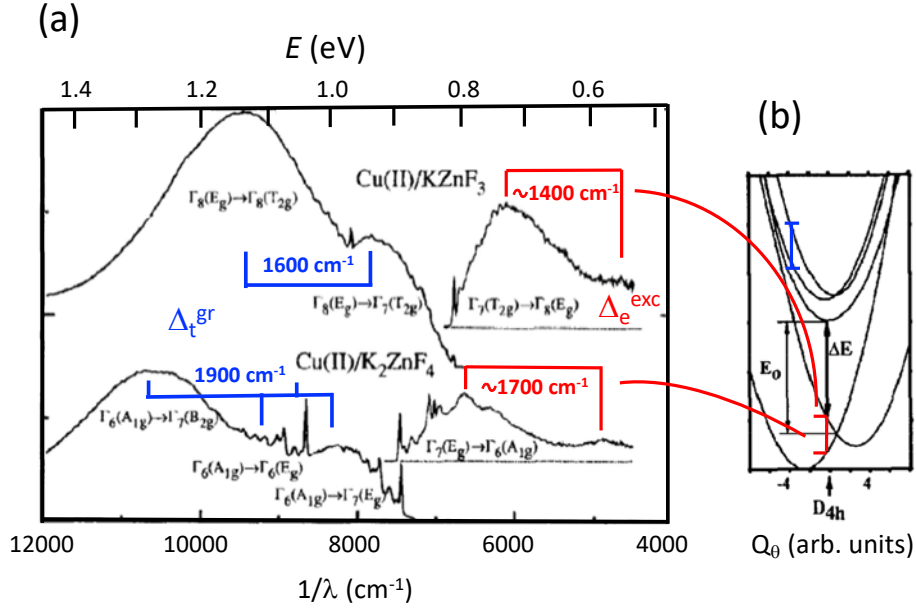


Figure S1

(a) Low-temperature emission and excitation spectra of $\text{KZnF}_3:\text{Cu}^{2+}$ and $\text{K}_2\text{ZnF}_4:\text{Cu}^{2+}$ showing a double-band structure associated mainly with the splitting of the parent octahedral e_g (red) and t_{2g} (blue) orbitals, Δ_e and Δ_t caused by the low-symmetry JT distortion of the CuF_6 polyhedron. Note that the splitting in excitation corresponds to the splitting of the octahedral $^2T_{2g}$ excited state at the equilibrium geometry of the 2E_g ground state distorted by the Jahn-Teller effect ($E \otimes e$) and crystal anisotropy (in K_2ZnF_4): Δ_t^{gs} . In the emission spectra, the observed double band structure corresponds to transitions from the $^2T_{2g}$ excited state to the octahedral 2E_g ground state split by the low-symmetry crystal field at the equilibrium geometry of the excited state ($E \otimes t$): Δ_e^{es} . It must be noted that observed emission splitting in emission, named Δ_e^{es} , corresponds to the tetragonal splitting of the octahedral e_g orbitals at the excited-state equilibrium geometry. It is different from the splitting $\Delta_e = \Delta_e^{\text{gs}}$ observed in absorption as it does really correspond to the e_g splitting at the ground-state equilibrium geometry. Indeed, Δ_e^{gs} is missed^{1,2} due to experimental difficulties and limitations of signal detection in the long wavelength spectral range.

(b) Q_θ -mode configuration energy diagram illustrating the ground and excited states splittings responsible for the emission and excitation band structure. Note that Δ_e^{es} is slightly smaller than Δ_t^{gs} in spite of $\Delta_e^{\text{es}} > \Delta_t^{\text{gs}}$ ($\Delta_e^{\text{es}} = 3.1 \Delta_t^{\text{gs}}$ from Fig. 4 of the manuscript).

Adapted from Figure 1 of reference² with permission of Elsevier Publishing, 1989.

2. Electron Paramagnetic Resonance of $(\text{CH}_3\text{NH}_3)_2\text{CdCl}_4$: Cu^{2+} : temperature dependence of the gyromagnetic principal values.

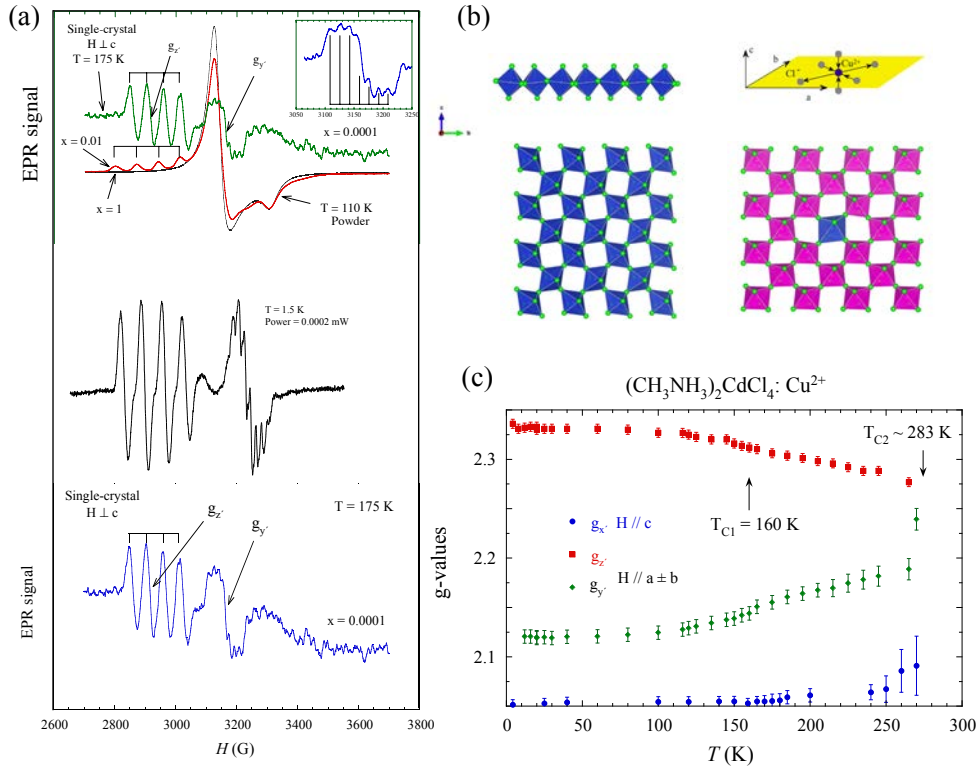


Figure S2

EPR spectra of Cu^{2+} introduced as a substitutional impurity in the layer perovskite $(\text{CH}_3\text{NH}_3)_2\text{CdCl}_4$.

(a) The plot shows the Cu^{2+} concentration dependence of single crystals of $(\text{CH}_3\text{NH}_3)_2(\text{Cd}_{1-x}\text{Cu}_x)\text{Cl}_4$ for $x = 0.0001, 0.01$, and 1 (pure copper crystal) at 175 K (single crystal) and 110 K (powder). Note that the spectra for the pure copper sample were obtained in single crystal and powder for comparison purposes. Besides the characteristic exchange averaged EPR spectra for $x = 1$, the characteristic signal of isolated Cu^{2+} become visible on decreasing the Cu^{2+} concentration. Only Cu^{2+} monomers are observed for $x = 0.0001$, while a mixture of monomers and aggregated phases are observed for intermediate concentrations. The lower the Cu^{2+} concentration the higher monomer to aggregate ratio. The monomer EPR spectrum ($x = 0.0001$) with the magnetic field applied perpendicular to the a, b plane of the layered perovskite is shown at 175 K (Tetragonal $(P4_2/n\text{cm})$) and 15 K (Monoclinic $P2_1/b$).³⁻⁴ The local gyromagnetic values g_x and g_y are indicated. The in-plane elongated Cu-Cl bond is along z , while the two short Cu-Cl bonds are along y (nearly in a, b plane) and x (nearly perpendicular to plane, or parallel to c).

(b) Schematic representation of the Cu^{2+} local structure in pure $(\text{CH}_3\text{NH}_3)_2\text{CuCl}_4$ (left) and $(\text{CH}_3\text{NH}_3)_2\text{CdCl}_4$: Cu^{2+} (right). Note that the impurity system resembles the antiferrodistortive structure exhibited by the pure crystal but monomer displays a dynamical Jahn-Teller distortion among these two locally elongated coordination geometries.

(c) Variation of the principal gyromagnetic values of Cu^{2+} monomer with temperature. It must be noted that in the time scale of GHz radiation (10^{-9} s) the equilibrium geometry of Cu^{2+} corresponds to a rhombic local coordination geometry with the two nearly in-plane elongated Cu-Cl bonds with $g_z = 2.33$, and the two short Cu-Cl bonds associated with $g_y = 2.12$ (nearly in plane larger bond) and $g_x = 2.055$ (nearly perpendicular to the crystal layer). The g values change with temperature according to a dynamically-induced geometrical exchange between the two in-plane elongated geometries. The in-plane g_y values evolve up to collapse to an average value $g_{z,y} = 2.24$, while g_x varies from 2.055 to 2.09 at high temperature, reflecting the actual nearly tetragonal symmetry at Cd^{2+} site. Interestingly, the variation of the principal g -values with the temperature, which are mainly governed by the energy barrier associated with the two in-plane elongated dynamical configurations, is affected by the structural phase transition sequence of the host crystal $(\text{CH}_3\text{NH}_3)_2\text{CdCl}_4$. This sequence favors the trend to a dynamically averaged configuration towards higher temperature phases. The energy barrier tends to decrease along the phase transition sequence: Monoclinic ($P2_1/b$) – 160 K – Tetragonal ($P6_2/n\text{cm}$) – 283 K – Orthorhombic ($Abma$).³⁻⁶

3. Electron Paramagnetic Resonance of $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4\text{:Cu}^{2+}$: temperature dependence of the gyromagnetic principal values.

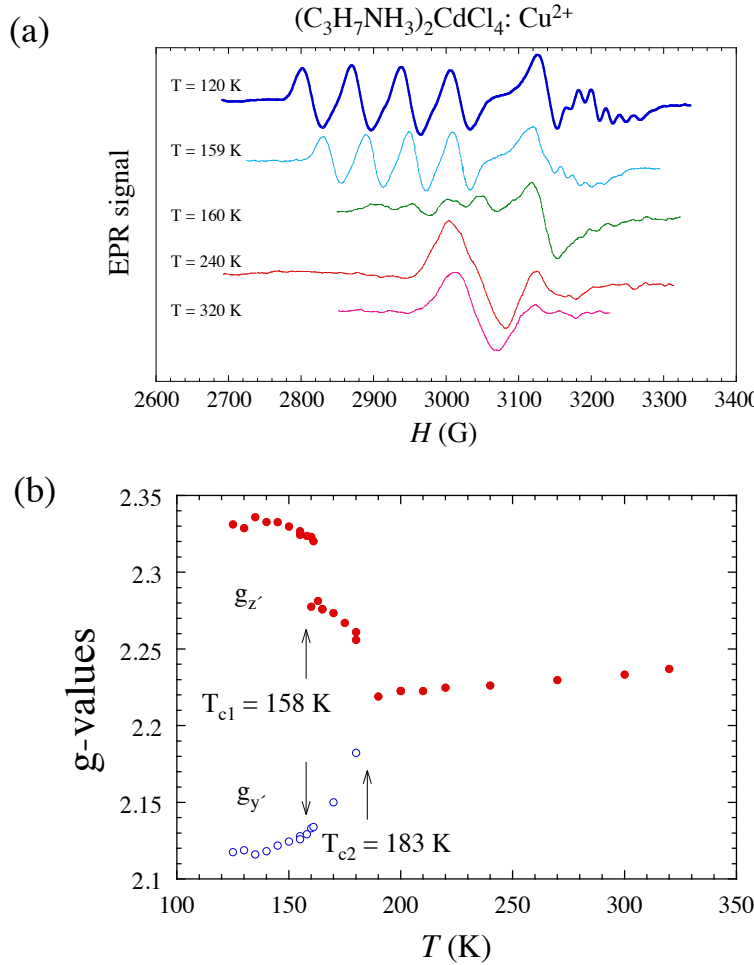


Figure S3

(a) EPR spectra and (b) gyromagnetic values of Cu^{2+} introduced as a substitutional impurity in the layer perovskite $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4$ as a function of temperature.

EPR spectra of diluted $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4\text{:Cu}^{2+}$ taken with the magnetic field perpendicular to the a,b plane of the layered perovskite (Orthorhombic $Pbca$ at 120 K)⁷ at different temperatures between 120 K and 320 K. Note that the low temperature spectrum exhibits the hyperfine and superhyperfine structures characteristic of a Cu^{2+} monomer hexacoordinated by six Cl^- with two nearly in-plane elongated Cu-Cl bonds ($g_z = 2.33$) and two short Cu-Cl bonds, one nearly in-plane ($g_y = 2.12$) and the other one nearly perpendicular to plane (along c). Interestingly, the evolution of the EPR spectra with temperature renders a nice example of a 2D dynamical Jahn-Teller effect between the two equivalent in-plane elongated Cu^{2+} geometries which are stabilized when Cu^{2+} replaces Cd^{2+} at the compressed Cd^{2+} site (the four long Cd-Cl bonds lie in the a,b plane while the two sort Cd-Cl bonds correspond to terminal ligands aligned perpendicular to it—nearly along c)⁷⁻¹⁰.

Similar to the isomorphous $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4$, the evolution of the g values with temperature reflects the 2D dynamical Jahn-Teller effect tending to converge to an average g -value when the thermally activated exchange rate is shorter than the time scale of the microwave radiation (X-band). Nevertheless, the energy barrier associated with the dynamical exchange for Cu^{2+} in $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4$ is smaller than those attained in $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4$ where averaged spectra are observed above 283 K. In present $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4\text{:Cu}^{2+}$, we do also observe an abrupt reduction of g_z and a concomitant jump of g_y at the structural phase transition of the host: Orthorhombic ($Pbca$) – 158 K – Orthorhombic incommensurate with $\mathbf{q} \approx 0.42\mathbf{b}^*$ ($Abma$)^{7,11,12} and a complete collapse at the structural phase transition to the Orthorhombic commensurate ($Abma$) phase at 183 K.^{11,12} Given that transition temperatures in $(\text{CH}_3\text{NH}_3)_2\text{CdCl}_4$ are lower than corresponding transition temperatures in $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4$ we conclude that energy barriers for dynamical exchange are reduced in the propylammonium crystal with respect to methylammonium one.

4. Electron Paramagnetic Resonance of $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CdCl}_4$: Cu^{2+} : temperature dependence of the gyromagnetic principal values and ferromagnetic resonance

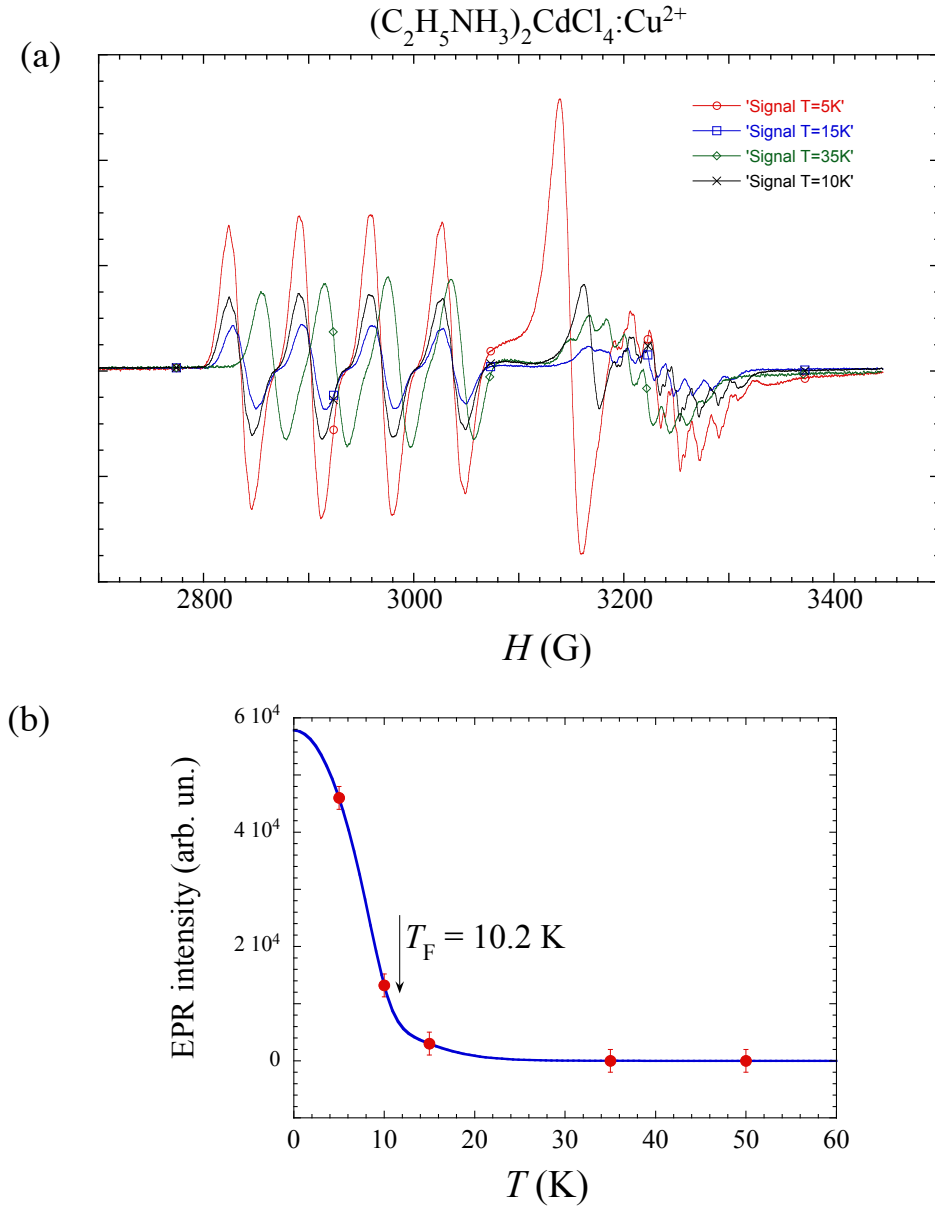


Figure S4

Structural characterization of Cu^{2+} introduced as a substitutional impurity in the layer perovskite $\text{C}_2\text{H}_5\text{NH}_3\text{CdCl}_4$ by means of EPR.

EPR spectra of diluted $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CdCl}_4:\text{Cu}^{2+}$ taken with the magnetic field perpendicular to the a,b plane of the layered perovskite (Monoclinic $B2_1/a$)¹³ at different temperatures between 5 K and 35 K (a). This Monoclinic phase is stable up to 114 K.¹³⁻¹⁶ Note that similar to $(\text{CH}_3\text{NH}_3)_2\text{CdCl}_4:\text{Cu}^{2+}$ and $(\text{C}_3\text{H}_7\text{NH}_3)_2\text{CdCl}_4:\text{Cu}^{2+}$, the low temperature spectrum of $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CdCl}_4:\text{Cu}^{2+}$ exhibits the hyperfine and superhyperfine structures characteristic of a Cu^{2+} monomer hexacoordinated by six Cl^- with two nearly in-plane elongated Cu-Cl bonds ($g_z = 2.33$) and two short Cu-Cl bonds, one nearly in-plane ($g_y = 2.12$) and the other one nearly perpendicular to the crystal layer. Besides the 2D dynamical Jahn-Teller effect at higher temperatures, a relative intense single EPR peak is observed below 10 K (b). We associate this peak with a ferromagnetic resonance coming from marginal small $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CuCl}_4$ aggregates. This interpretation is based on the fact that this kind of aggregates are formed upon increasing the Cu^{2+} concentration, the exchange-narrowed gyromagnetic ratio (the ferromagnetic resonance magnetic field) increases (decreases) with decreasing temperature below the Curie temperature, which is $T_F = 10.2$ K for bulk $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CuCl}_4$.^{17,18} The line intensity falls down at the Curie temperature. The variation of $I(T)$ suggest that the actual Curie temperature might be a few degrees higher than bulk probably due to the small size of aggregates or the slightly lower exchange interactions provoked by the larger $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CdCl}_4$ lattice in comparison to $(\text{C}_2\text{H}_5\text{NH}_3)_2\text{CuCl}_4$.

5. Jahn-Teller and crystal anisotropy (stress) effects on the local structure of MX_6 (M : Cu^{2+} , Mn^{3+} ; X : F, Cl) in ABX_3 : M perovskite-type structures.

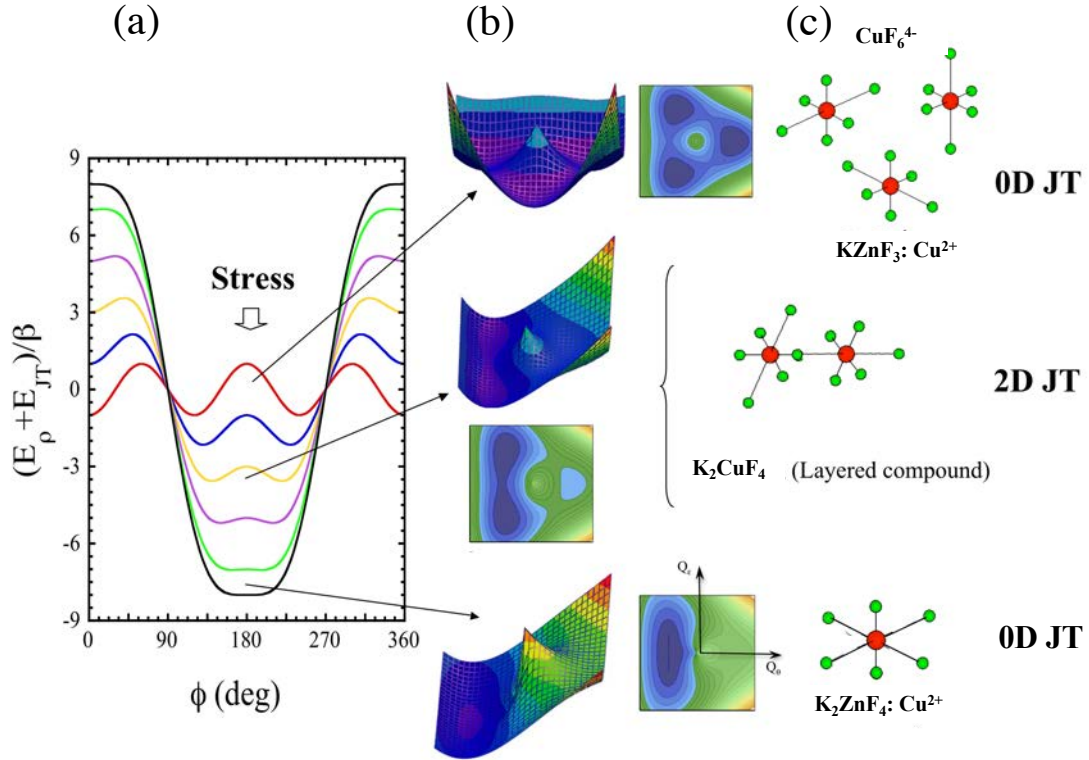


Figure S5

(a) Effect of crystal anisotropy represented through an effective axial stress in the ground-state energy in (Q_θ, Q_ϵ) -space given by the $E \otimes e$ Jahn-Teller theory.^{10,18-20} (b) (Q_θ, Q_ϵ) -space ground-state energy surface. Note the evolution from elongated-to-compressed coordination geometry upon increasing the axial stress. The collapse into the compressed geometry takes place at $S_{\text{crit}} = 9\beta$. The parameter β (> 0 for elongated geometry minima) contains anharmonic and second-order JT interactions yielding warping of the Mexican-hat-type energy surface in (Q_θ, Q_ϵ) -space. 2β corresponds to the energy barrier for jumping among energy minima at $S = 0$. (c) Structures illustrate the CuF_6^{4-} equilibrium geometries predicted by the perturbed JT model.¹⁸⁻²⁰ The stress parameter is introduced in the Jahn-Teller model as $S = \pm A_e \rho_0$, with A_e and ρ_0 being the electron-lattice coupling constant associated with the $E \otimes e$ Jahn-Teller effect and the low-symmetry coordinate at the host site with plus or minus depending on whether the type of tetragonal distortion is compressed (compressive stress) or elongated (tensile stress), respectively.²⁰

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